

Report Sheet:**Arsenic Toxic Levels****CHEM 110 / 154****CAPILANO
UNIVERSITY**

Department of Chemistry

LAST NAME: _____ SEC # _____ LOCKER # _____

FIRST NAME: _____ DATE: _____

Record all data at the appropriate location on this report sheet. Do not drop "leading zeroes".

Raw Data

Report all masses as given in the Virtual Lab software (typically 6 significant figures).

Volume of AgNO ₃ added (mL) <i>only add as needed</i>	Mass Ag ₃ AsO ₄ obtained (g)		
	0.01 M Na ₃ AsO ₄ Standard	Sample #1 analyte (sample <u>A</u> on graph)	Sample #2 analyte (sample <u>C</u> on graph)
1.00			
2.00			
3.00			
4.00			
5.00			
6.00			

Samples B and D

As noted in the lab manual (Treatment of Results section) Samples B and D were collected downstream from Samples A and C, respectively, and are given in the online Raw Data as a percent of A and C. Record below your uniquely-assigned percents and then use these, along with your mass values for A and C, to calculate the mass of Ag₃AsO₄ obtained for Samples B and D. Report to 6 significant figures.

Sample	Percent of:	Mass Ag ₃ AsO ₄ obtained (g)
# B	_____ of Sample A	=
# D	_____ of Sample C	=

Calculations:**Part One**

From the mass, calculate the **moles of Ag_3AsO_4 precipitated** in the flask of Na_3AsO_4 standard. Give your answer to 5 significant figures:

_____ mol Ag_3AsO_4

Using the volume and concentration of the Na_3AsO_4 standard, **calculate the number of moles of arsenate, AsO_4^{3-} , initially present** in the flask. Give your answer to 5 significant figures:

_____ mol AsO_4^{3-}

Calculate, to 2 decimal places, the **percent of AsO_4^{3-}** that was precipitated out of solution:

_____ %

Is gravimetric analysis of AsO_4^{3-} with AgNO_3 a reliable method? Refer to the Introduction in the lab manual. Checkmark your choice: ☐ **YES** ☐ **NO**

Calculations:**Part Two**

Calculate, to 0 decimal places, the Arsenic Levels for each sample, as discussed in the lab manual. Show the calculation for Sample A on the next page. Also, record below your uniquely-assigned “distance-from-mine” values from your online version of the Raw Data. This table below will be needed to create your graph.

sample	Distance from Mine (km) *	Calculated As Level (mg / L)
A	0	
B		
C		
D		

* from website

Arsenic Level Calculation for Sample A: Show all your work; final answer to 0 decimals.

_____ mg As / L

Calculations:

using the Graph

Record here the equation for the line-of-best-fit. Give the **slope to 2 decimal places** and the **y-intercept to 0 decimal places**:

$$y = \frac{\text{_____}}{\text{slope}} x + \frac{\text{_____}}{\text{intercept}}$$

Record the statistical R^2 value: _____ (a value of 1.0 means a perfect fit)

Checkmark the software used to prepare the graph: ☐ MS-Excel ☐ Google Sheets

If your Arsenic Level for Sample D is above the MCL of 0.01 mg / L, use the equation of the line to calculate the “**minimum safe distance**” from the mine, in units of kilometers, to 1 decimal.

_____ km

A cabin on private property is located 1 km downstream from where Sample D was collected. You can't access the property to take a reading from the river, but the home-owner should be notified if they are in danger. Calculate the predicted arsenic level, to 0 decimal places.

Distance of cabin from mine using my assigned Sample D data: _____ km

Predicted Arsenic Level: _____ mg As / L.

Please staple your graph to the back of this lab report sheet before coming to the lab.